



## Study of Surface Water and Soil Quality Affected by Heavy Metals of Pabna Sadar

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### Abstract

Concentrations of heavy metals such as As, Cd, Co, Cr, Ni, Fe, Zn, Pb, Hg and Cu in surface water and soils were determined by using Atomic Absorption Spectrophotometer (AAS) using standard analytical methods. The physicochemical and anionic properties of the surface water were also determined to assess the pollution level of this area. The results reveal that although few parameters of the surface water are within the standard value but most of the physicochemical and anionic parameters such as BOD, COD, F<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, NH<sub>3</sub> and Pb of the surface water are significantly higher than the standard for drinking water. The result suggests that the surface water of this area is polluted by various pollutants and harmful for drinking and recreational purposes. Most of the soil contains much higher values of heavy metals like Hg, Mn, Co and Zn than world wide average soil values (0.05, 270.00, 5.50, 45.00 ppm respectively). Therefore, it might have considerable negative effects on the soil quality, agricultural crops of the area and thus harmful for human health. Their higher concentration suggests that they resulted due to both natural origin and anthropogenic activities like, application of pesticides, mineral fertilization and industrial discharge.

**Keywords:** Heavy metals, Physicochemical properties, Surface water soil quality, Human health.

### Introduction

In developing tropical countries like Bangladesh metal contamination has a significant impact on environmental health. Surface water bodies through out Bangladesh are also subject to potential water-quality hazards associated with metals due to both intense chemical weathering and anthropogenic activities (Islam *et al.*, 2000). The major problems associated with excessive release of metals into the environment are that metals neither biodegradable, nor they are eliminated by incineration processes. These elements tend to persistent pollutants, and can accumulate in ecosystems and food chains. In addition, each metal has a specific chemical form (speciation) which determines its solubility in water, and consequently its ability to incorporate into biological systems. Organometallic compounds are able to pass through biological membranes because of their high degree of lipophilicity. Heavy metals like Cd, Cu, Ni, Hg, Pb, and Zn represent potential hazards to plants and animals (Greenland *et al.*, 1981).

Modern civilization is now dependent on the large-scale use of a wide range of metals and most of them are naturally present only at trace levels in the hydrosphere (biosphere) (Chow, 1968). A major route by which trace elements are dispersed in the hydrosphere (biosphere) is associated with the disposal of metal containing industrial effluents.

Industrial effluents discharged directly into the sea or into waterways or sewers, but whatever the disposal route, they constitute an important source of metals contaminating the environment. (Park, 1972, Chunye *et al.*, 2008). Behaviour of trace metals in soil depends upon complex reactions among micro and macro cations and anions and different components of various soil phases, solid, aqueous and gaseous. Soils of several regions of the world have been subjected and will be in the future to mineral fertilization, pesticide application, and industrial pollution and, waste disposal. All these activities will affect both chemical and physical soil properties & will lead to changes in the behaviour of trace element in soils (Salmons *et al.*, 1995, Panichayapichet *et al.*, 2007).

Reliable analytical methods play an extremely important role in monitoring the content of heavy metals in soil. Recently, atomic absorption spectrophotometric methods, anodic and cathodic stripping voltametric method, ICP spectrometry, Instrumental neutron activation analysis and X-ray fluorescence spectrometry for the determination of soils have been used (Brenner, 1998). No systematic work has been carried out to date to assess the possibility of metal contamination of surface waters in the northern region of Bangladesh. There are many industries (BSCIC, pharmaceutical, toiletries)

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situated in Pabna Sadar area, so many pollutants are continuously discharging to the surrounding areas. People who live surrounding area utilizing surface water for their house hold work, bathing, irrigation, fish culture and other necessary works.

Although the level of heavy metals in few selected areas of Bangladesh have been found in the literatures (Chamon *et al.*, 2009, Shamshad *et al.*, 2005, Sultana *et al.*, 2009 & 2003, Ali *et al.* 1998, Kashem *et al.*, 1998, Alam *et al.* 2002, Mahfuz *et al.*, 2004), but no systematic work about it has done at Pabna Sadar. So it is necessary to assess the soil and water quality of this area to concern people as well as to protect environment. So the purpose of this study was to determine the concentration of physicochemical properties and heavy metals in surface water and soil in order to asses the extent of pollution level of this area by comparing the surface water data of the study area with DoE standard for drinking water. Further the data of the studied soil was compared with average data for world wide soil and other areas soil of Bangladesh.

**Materials and Methods**

**Study Area**

Pabna Sadar is an upazila of Pabna district in the division of Rajshahi, Bangladesh (Fig.1). It is located at latitude

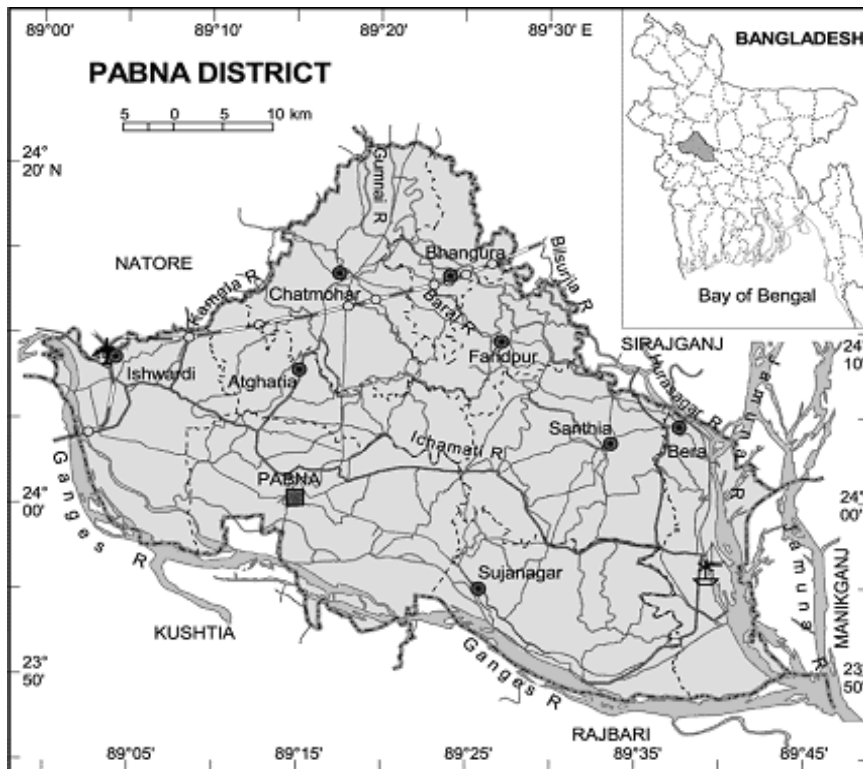
24.00°N and longitude 89.25°E, has a tropical environment. It has an area of 443.90 sq km, which is bounded by Atgharia and Ishwardi upazilas on the north, Kushtia sadar, Kumarkhali, Khoksa and Pangsha upazilas on the south, Santhia and Sujanagar upazilas on the east, Ishwardi upazila on the west. Main rivers are Ganges and Ichamati and noted depressions are Sonapatil Beel and Tara pasha Canal.

**Sample collection**

In this study, the sampling sites are located inside Pabna Sadar Upazila. Twenty (20) samples were collected from the study area. Among them ten (10) surface water samples were collected in plastic bottles from ponds, lakes and rivers and 10(ten) soil samples(0- 15 cm) were collected randomly in the vicinity of the surface-water sampling sites. Samples were collected by proper sampling procedure in rainy season (May, 2007) and analysis of the samples was performed around two months by using standard analytical methods. The locations of the sampling sites and sources with ID numbers are shown in Table Ia and Ib.

**Sample Analysis**

The physicochemical parameters, anion concentrations and heavy metal concentrations of surface water (lakes, ponds



**Fig. 1: The map of the study area, Pabna District.**

and rivers) samples were analyzed by different standard methods in the laboratory of analytical research division, BCSIR laboratories, Dhaka. Physicochemical parameters such as pH, total suspended solid (TSS), total dissolved solid (TDS), electrical conductivity (EC), turbidity, dissolved oxygen (DO), biochemical oxygen demand (BOD), chemical oxygen demand (COD) of the surface water samples were measured using various standard methods (APHA, 1976). A quantitative analysis of fluoride ( $F^-$ ), Chloride ( $Cl^-$ ), Bromide ( $Br^-$ ), Nitrate ( $NO_3^-$ ), Nitrite ( $NO_2^-$ ), and Sulfate ( $SO_4^{2-}$ ) of the surface water samples were determined by Ion Chromatography (IC). A quantitative analysis of Ammonia ( $NH_3$ ) and Phosphorus (P) of the surface water were determined by UV Spectrophotometer (Model- Lambda 25, Perkin Elmer) at 400nm wave length. Phosphate ( $PO_4^{3-}$ ) concentration of the surface water samples was determined from phosphorus (P) concentration using the factor  $(95/31=3.0645)$  (Sultana *et al.*, 2009). Pb, Zn, Cd and Cu - concentrations of the surface water were determined by Anodic Stripping Voltammetry (AVS) and Cr, Mn, Fe, Co, Ni,

As and Hg were determined by Atomic Absorption Spectroscopy (AAS). The soil samples were air dried, sieved (1mm) and after oven dried at 80°C, ground to powder. After acid digestion, the residue was dissolved in dilute acid solution to yield the sample solutions. Further a quantitative analysis of heavy metals such as Chromium (Cr), Manganese (Mn), Iron (Fe), Cobalt (Co), Nickel (Ni), Arsenic (As), Zinc (Zn), Cadmium (Cd), Mercury (Hg) and Lead (Pb) of the soil were determined by Atomic Absorption Flame Emission Spectrophotometer (AAS) Model: SHIMADZU, AA-6401F.

## Results and Discussion

### Concentrations of the physicochemical properties of the surface water samples

The concentration ranges of the physicochemical properties with the average value of the surface water of the study area are shown in Table II. The levels of pollution of the surface water were determined by comparing the observed values of

**Table Ia: Locations and sources of the sampling sites of the surface water samples with ID numbers**

Serial No.	Sampling sites	Sources	Sample ID
1	BSCIC (Bangladesh Small & Cottage Industries Corporation)	Pond	S-1
2	BSCIC (Bangladesh Small & Cottage Industries Corporation)	Pond	S-2
3	Pailanpur	Pond	S-3
4	Dilalpur	Pond	S-4
5	Dilalpur	Pond	S-5
6	Baiorampur	Lake	S-6
7	Dakshin Ramchandrapur(Padmar koal)	River	S-7
8	Bangla Bazar	River	S-8
9	Dakshin Ramchandrapur(Islampur)	Lake	S-9
10	Shinga	River	S-10

**Table Ib: Locations and sources of the sampling sites of the surface soil samples with ID numbers.**

Serial No.	Sampling sites	Types of Soil	Sample ID
1	BSCIC (Bangladesh Small & Cottage Industries Corporation)	Uncultivated	S-1
2	BSCIC (Bangladesh Small & Cottage Industries Corporation)	Uncultivated	S-2
3	Pailanpur	Uncultivated	S-3
4	Dilalpur	Uncultivated	S-4
5	Dilalpur	Uncultivated	S-5
6	Baiorampur	Irrigated	S-6
7	Dakshin Ramchandrapur(Padmar koal)	Uncultivated	S-7
8	Bangla Bazar	Irrigated	S-8
9	Dakshin Ramchandrapur(Islampur)	Uncultivated	S-9
10	Shinga	Uncultivated	S-10

the various parameters with the drinking water standards value recommended by DoE, Bangladesh (Huq, 2003).

Water pH influences the other properties of water body, activity of organisms, and potency of toxic substances present in the aquatic environment. The pH of the surface water samples was varied from 7.30 to 9.12. The average pH value was found 8.23. The electric conductivity (EC) is usually used for indicating the total concentration of the ionized constituents of water. The highest electric conductance was found at 1120  $\mu\text{s}/\text{cm}$  and the lowest electric conductance value observed was 333  $\mu\text{s}/\text{cm}$  and the average value was found 856  $\mu\text{s}/\text{cm}$ . TDS in water mainly consist of ammonia, nitrite, nitrate, phosphate, alkalis, some acids, sulphates, metallic ions etc (Moore *et al.*, 1960). The TDS values of the study area lies between 160.8 mg/L and the 624 mg/L. The average value was found 423.98 mg/L. Total suspended solids denote the suspended impurities present in the water. Measurement of suspended particulate matter is important, as they are responsible for pollutant transport in the aquatic environment. The TSS values of the study area lie between at 0.08 mg/L and the 92.42 mg/L. The average value was found 10.29 mg/L. The presence of oxygen in water is a positive sign of water but the absence of oxygen is a signal of severe pollution. The highest dissolved oxygen (DO) value of the surface water sample was found 5.6 mg/L and the lowest dissolved oxygen (DO) of the waste water observed was 3.1 mg/L and the average value was found 4.02 mg/L. Salinity is an important property of industrial and in natural water it is originally conceived as a measure of the mass of dissolved salts in a given mass of solution. The highest salinity value of the surface water was found 0.5 mg/L and the lowest salinity value observed was 0.1 mg/L. The average value was found 0.37 mg/L. Turbidity in water is caused by suspended and colloidal matter such as clay, silt, finely divided organic and inorganic matter, and plankton and other

microscopic organisms. The highest turbidity of the surface water was found 53 NTU and the lowest Turbidity value observed was 3.83 NTU. And the average value was found 16.05 NTU.

BOD is an index of the biodegradable organics present. The BOD values of the study area lie between at 1.24 mg/L and the 3.7 mg/L. The average value was found 2.27 mg/L which is ten times higher than the standard value. Chemical oxygen demand (COD) measures the oxygen required for the oxidation of mainly organic matter by a strong chemical oxidant. The COD of the surface water samples varied from 14.08 mg/L to 103.84 mg/L and the average value was found 37.13 mg/L. It is also ten times higher than the DoE standard. It indicated that the surface water of this area contain high content of biodegradable organic materials.

#### Concentrations of the anionic properties of the surface water samples

The concentration ranges of the physicochemical properties with the average value of the surface water of the study area are shown in Table III. The concentration of fluoride which is an important inorganic anion of water and effluents varied from 0.57 to 2.75 ppm. The average value of the  $\text{F}^-$  ion, of the surface water samples was found 2.13 ppm which is two times higher than DoE standard. Chloride ( $\text{Cl}^-$ ) is an important chemical parameter to determine water quality. The  $\text{Cl}^-$  conc. of the surface water samples varied from 48.91 to 203.46 ppm and the average value was found 161.20 ppm. The  $\text{NO}_3^-$  concentration of the surface water samples lies between 0.15 ppm and 3.89 ppm. The average value was found 1.09 ppm. The concentration of nitrate ( $\text{NO}_3^-$ ) which is one of the critical nutrients for the growth of algae and helps accelerating the eutrophication varied from 12.74 to 70.77 ppm and the average value was found 39.84 ppm. It is also 4 times higher than DoE standard. Sulfate is an important

**Table II: Descriptive statistics of the physicochemical parameters of the surface water samples**

Parameter	Maximum	Minimum	Mean	Std. Devi( $\pm$ )	DoE std. (ppm)
TSS (mg/L)	92.42	0.08	10.29	28.88	
TDS (mg/L)	624	160.8	423.98	132.98	1000
Turbidity (NTU)	53	3.83	16.05	15.04	10
DO (mg/L)	5.60	3.10	4.02	1.01	6
pH	9.12	7.30	8.33	0.58	6.5-8.5
EC ( $\mu\text{s}/\text{cm}$ )	1 120.00	333	856.00	244.99	
BOD (mg/L)	3.70	1.24	2.27	0.71	0.2
COD (mg/L)	103.84	14.08	37.13	26.25	4
Salinity(mg/L)	0.50	0.10	0.37	0.12	

anion imparting hardness to the water. The highest  $\text{SO}_4^{2-}$  of the surface water samples was found 26.26 ppm and the lowest was observed. And the average value was found 17.25 ppm. Phosphorus (Phosphate-phosphorus) is generally the limiting nutrient for algal growth and controls the primary productivity of water body. High levels of phosphate may originate from municipal wastewater discharges, since it is an important component of detergents. The concentration of  $\text{PO}_4^{3-}$  of the surface water samples varied from 0.72 to 10.03 ppm and the average value was found 4.99. The concentration of ammonia ( $\text{NH}_3$ ) of the surface water samples lie between 1.02 and 3.44 ppm and the average value was found 1.83 ppm. It is also 3 times higher than DoE standard.

Zn and Cd concentrations were found below the minimum detection limit.

Therefore, most of the parameters such BOD, COD,  $\text{F}^-$ ,  $\text{NO}_3^-$ ,  $\text{NH}_3$  and Pb of the surface water were found significantly higher than the DoE standard for drinking water. The result suggests that heavy metals and other pollutants pollute the surface water. So the surface water is harmful for drinking and recreational purposes.

#### Concentrations of the heavy metals of the soil samples

Total concentrations of the heavy metals in the soils of the various locations of the study area are shown in Fig 2. Further the range of the heavy metal concentrations with the average value of the soils of this area are shown in Table V.

**Table III : Descriptive statistics of the anionic parameters of the surface water samples**

Anion	Maximum	Minimum	Mean	Std. Devi(±)	DoE std. (ppm)
Fluoride ( $\text{F}^-$ )	12.75	0.57	2.13	3.74	1
Chloride ( $\text{Cl}^-$ )	203.46	48.91	161.20	57.24	150-600
Nitrite ( $\text{NO}_2^-$ )	3.89	0.15	1.09	1.23	
Nitrate ( $\text{NO}_3^-$ )	70.77	12.74	39.84	21.11	<1
Sulfate ( $\text{SO}_4^{2-}$ )	26.26	5.52	17.25	7.96	400
Phosphate ( $\text{PO}_4^{3-}$ )	10.03	0.72	4.99	2.83	6
Ammonia ( $\text{NH}_3$ )	3.44	1.02	1.83	0.72	0.5

#### Concentrations of the heavy metals of the surface water samples

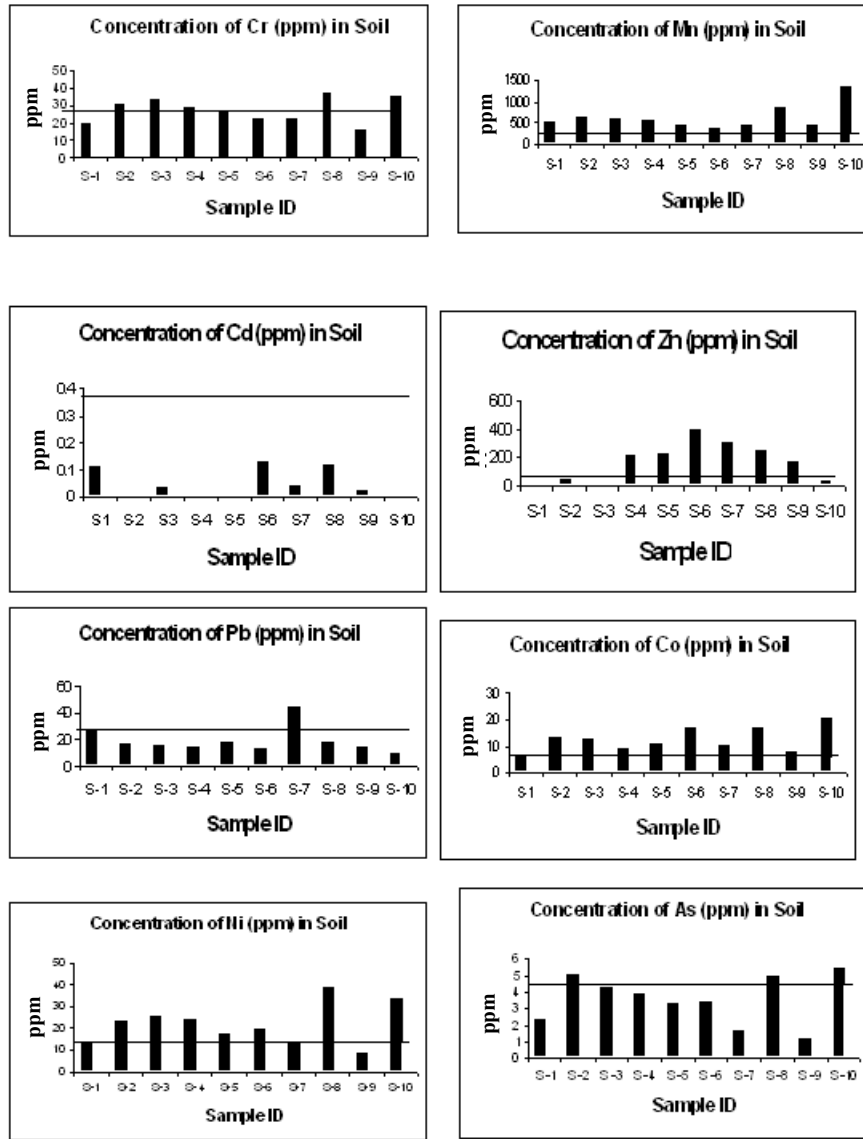
The concentration ranges of the heavy metals with the average value of the surface water of the study area are shown in Table IV. The average concentrations of Cr, Mn, Fe, Ni, As, Hg and Pb of the surface water samples were found 0.0050 ppm, 0.0961 ppm, 0.265 ppm, 0.0097 ppm, 0.0078 ppm, 0.00032 ppm and 0.0626 ppm. The concentration of Pb in the surface water is found slightly higher than the DoE water standard. In the collected surface water samples Cu,

#### Comparison of the present data with the world average soil, soil of Rajarampur and soil of Savar

In this study, the heavy metal concentrations of the soils of the study area are compared with the average concentrations of heavy metals for worldwide soils (Alina K.P.,1992), the uncultivated soils of industrial area of Savar (Sultana *et al.* 2003) and the soils of the same geological region, Rajarampur (Islam *et al.*, 2000) of the present studied soil Pabna Sadar (Table VI). The average concentration of As in the studied soil samples was found 3.56 ppm, which is

**Table IV: Descriptive statistics of the heavy metal concentration of the surface water samples**

Metals	Maximum	Minimum	Mean	Std. Devi(±)	DoE std. (ppm)
Cr	0.0134	0.0002	0.0050	0.0040	0.05
Mn	0.1917	0.012	0.0961	0.0543	0.1
Fe	0.6551	0.1233	0.2659	0.1588	0.3-1
Co	0.0388	0.0014	0.0158	0.0153	
Ni	0.0301	0.002	0.0097	0.0102	1
As	0.0254	0.0006	0.0078	0.0076	0.05
Hg	0.0007	0.00001	0.0003	0.0002	0.01
Pb	0.2319	0.0091	0.0626	0.0950	0.05



**Fig. 2:** The concentrations of heavy metals in the soils of the different locations of Pabna sadar. The solid line represents the world average value (Alima *et al.*, 1992).

comparable to the data for worldwide soils, the uncultivated soils of industrial area of Savar and the soils of Rajarampur.

The average concentration of Cd was found 0.07 ppm which is lower than the data for world wide soils and uncultivated soils of Savar. The average concentration of Cr of the soil samples was found 27.10 ppm which is lower than the data for world wide soils, uncultivated soils of Savar and comparable to the data for the soils of Rajarampur. The average concentration of Ni of the soil samples was found 21.77 ppm which is comparable to the data for worldwide soils, the uncultivated soils of industrial area. The average concentration of Pb of the soils was found 18.92 ppm which is compa

able to the data of for worldwide soils, the uncultivated soils industrial area, Savar and the soils of Rajarampur. The above results suggest that the soils of this area accumulated the heavy metals like As, Cd, Cr, Ni and Pb from the parent rocks during the soils formation process.

On the other hand, in the soil of the study area, the average value of Hg was found 1.84 ppm which is around 36 times higher values than the data for worldwide soils. The average concentration of Mn in the studied soil was found 628.91 ppm, which is around two times higher than the data for worldwide soils. The average concentration of Co in the studied soils was found 12.55 ppm which is also two times

**Table V: Descriptive statistics of the heavy metal concentrations of the soil samples**

Metals	Maximum	Minimum	Mean	Std. Devi(±)
Cr	37.42	16.56	27.10	7.16
Mn	1369.99	389.52	628.91	292.65
Fe	47584.5	11178.7	25544.53	12410.48
Co	20.81	6.95	12.55	4.37
Ni	38.98	8.11	21.77	9.27
Zn	395.09	0.30	165.04	136.73
As	5.40	1.19	3.56	1.46
Cd	0.13	0.02	0.07	0.05
Hg	3.21	0.82	1.84	0.62
Pb	44.49	9.33	18.92	10.04

higher than the data for world wide soils. Further, the average concentration of Zn was found 165.04 ppm and this value is around five times higher than the data of for world-wide soils, the uncultivated soils of industrial area Savar and the soils of Rajarampur. The higher content of heavy metals like Hg, Mn, Co and Zn in the soils of the study region suggests that inputs these from anthropogenic sources through irrigation, mineral fertilization, and industrial pollution, application of pesticides or waste disposal.

The above results suggest that, most of the soil contain much higher values of heavy metals than world wide average soil value and have considerable negative effects on the soil quality and agricultural crops of the area and thus harmful for human health.

## Conclusion

The present study was carried out in order to assess the environmental impact of heavy metals on the surface water and soil quality of the Pabna Shadar. A total of 10 surface water samples and 10 soil samples were collected from the surrounding area of Pabna Shadar. Physicochemical parameters like pH, TSS, TDS, EC, turbidity, DO, BOD, COD; anionic pollutants ( $F^-$ ,  $Cl^-$ ,  $Br^-$ ,  $NO_2^-$ ,  $NO_3^-$ ,  $SO_4^{2-}$  and  $PO_4^{3-}$ ) and heavy metal (Cr, Mn, Fe, Co, Ni, Cu, Zn, As, Cd, Hg and Pb) concentrations of the collected samples were investigated. The concentrations of the surface water samples were compared with the drinking water standard value recommended by DoE in order to assess the quality of the surface water samples of the study area. The concentrations of the soil samples were compared with the average value of worldwide soil and soils of other areas of Bangladesh. The results reveal that although few parameters of the surface water were found within the standard value of DoE, most of the physicochemical and anionic parameters such as BOD, COD,  $F^-$ , Mn,  $NH_3$  and Pb of the surface water were found significantly higher than the DoE standard for drinking water. The result suggests that the surface water of this area is polluted by various pollutants and harmful for drinking and recreational purposes. Further, most of the soil contain much higher values of heavy metals than world wide average value and have considerable negative effects on the soil quality, agricultural crops of the area and thus harmful for human health. The higher content of heavy metals like Hg, Mn, Co and Zn in the soils of the study region suggests that they resulted due to natural origin and anthropogenic activities like irrigation, mineral fertilization, and application of pesticides, industrial discharge.

**Table VI: Comparison of the heavy metal concentrations of the soil of the study area with heavy metal concentrations of other soils**

Heavy metals	Mean value of the soil of Pabna Sadar (ppm) <sup>a</sup>	Average Value of the world soil (ppm)	Mean value of the soil of Savar (ppm) <sup>b</sup>	Mean value of the soil of Rajarampur (ppm) <sup>c</sup>
As	3.56	4.40	7.00	<10
Cd	0.07	0.37	0.20	
Co	12.55	5.50	12.00	
Cr	27.10	47.00	77.50	33.54
Hg	1.84	0.05		
Mn	628.91	270.00		
Ni	21.77	13.00	33.00	18.33
Pb	18.92	22.00	20.50	16.18
Zn	165.04	45.00	60.60	57.72

a. Analyzed by Alina *et al.*, (1992).

b Analyzed by Sultana *et al.*, (2003)

c. Analyzed by Islam *et al.*, (2000).

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